41256

B. Sc. (Hons.) Mathematics 4th Semester Examination – May, 2019

CHEMISTRY-IV

Paper: BHM-245 Opt.-ii

Time: Three hours J

[Maximum Marks : 60

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

Note: Attempt five questions in all, selecting not more than two questions from each Section.

SECTION - I

- (a) Define auto-ionisation. Write self ionization reaction of: H₂O, NH₃, SO₂.
 - (b) Explain the nature of SOCl₂, Na₂SO₃ as acid /base in liquid SO₂ as a solvent.
 - (c) Write down the general characteristics of Lanthanides.
- (a) Give points of similarities between lanthanides and actinides.

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| | (b) | Explain Bronsted-Lowry Acid & Base concept with example. |
|----|-----|---|
| | (c) | Why Actinides have more tendency to form complexes than actinides. |
| 3. | (a) | Define Lewis acid and Lewis base with an example. |
| | (b) | Select following as acid and base with reason: AlCl ₃ , H ₂ O, NH ₃ , HCl. 4 |
| | (c) | Explain Hard & Soft Acid & Base concept with example. |
| | (d) | Name the Lanthanide which is radioactive. |
| | (e) | Define Transuranic elements with example. |
| | | SECTION - II |
| | | SECTION - II |
| 4. | (a) | Explain 3rd law of thermodynamics along with their application. |
| | (b) | Derive Clausius-Claypeyron equation. 5 |
| | (c) | Write a note on: |
| | | (i) Weston standard cell |
| | | (ii) Standard Hydrogen Electrode |
| 5. | (a) | Explain the spontaneity of the process in terms of: |
| | | (i) Free energy |
| | | (ii) Enthalpy |
| | | (iii) Entropy |
| | (b) | Define: 2 |
| | - | (i) Standard electrode potential |
| | | (ii) Activity |
| | | (2) |

| 8. | (a) | What is the structure of: | 2 |
|----|-------------|---|----------------|
| | | (i) Formalin | |
| | | (ii) Acetophenone | |
| | (h) | Explain reactivity order among aldehydes a ketones towards nucleophilic addition reaction the basis of: | and on 4 |
| | | (i) Inductive effect | |
| | | (ii) Steric effect. | |
| | (c) | Explain the mechanism of following reaction: | 6 |
| ! | | (i) Aldol condensation | |
| | | (ii) Wittig reaction | |
| | (a) | Write the following reactions: | 8 |
| | | (i) Gabriel Phthalimide reaction | |
| | | (ii) Hofmann bromamide reaction | |
| | | (iii) Benzoin condensation | |
| | | (iv) Diazotisation reaction | |
| | (b) | Do the following conversions: | 4 |
| | | (i) Aniline into Acetanilide | |
| | | (ii) Ethanamine into Ethyl-isocyanide | |
| | | (iii) Ethanal into Acetone | |
| | | (iv) Propanone into Iodoform | |
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